

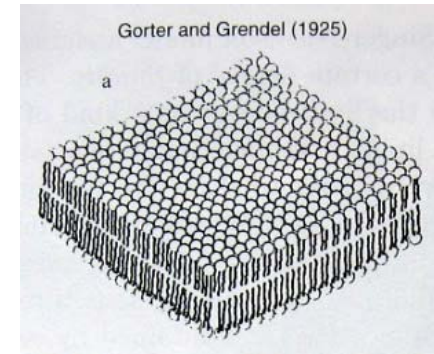
Lipids, lipid mixtures, and membranes

Voet & Voet Chapter 12

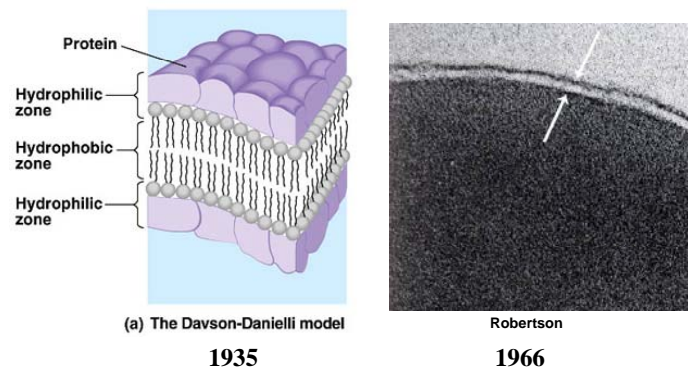
1, 2, 3A to C

Van Meer et al. 2008. Nature Reviews Mol.Cell.Bio. 9, 112

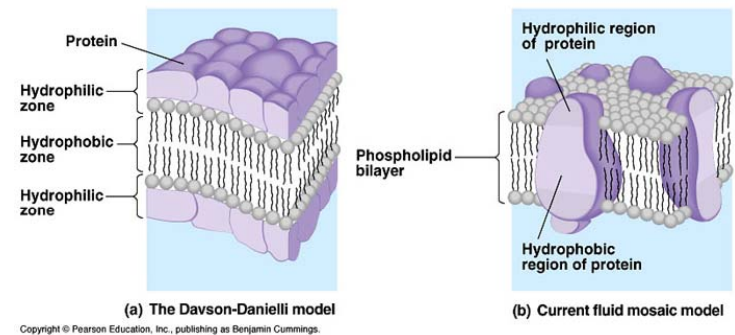
Early model of the erythrocyte membrane



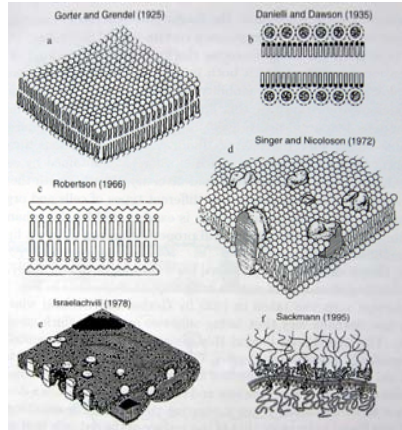
The structure of the cellular membrane



Singer and Nicolson 1972

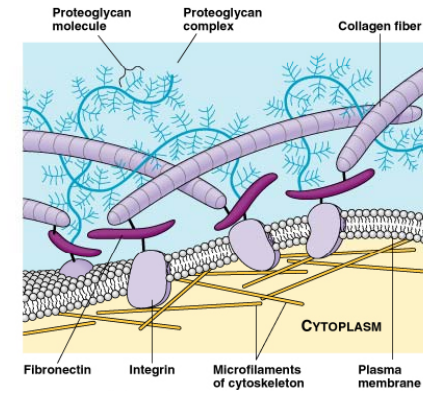


Historic picture gallery of membrane modelling

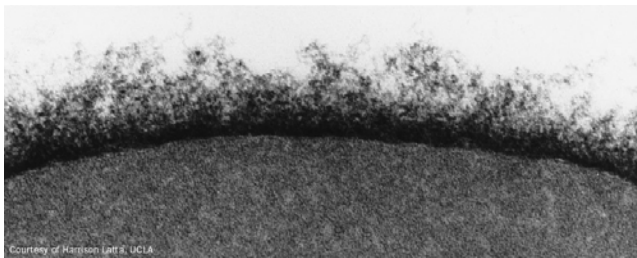


from the book
Life - as a matter of fat
O. Mouritsen
Springer 2005

Current model of the cellular membrane



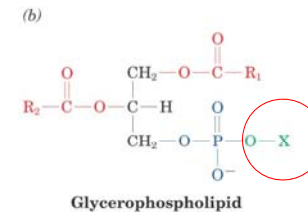
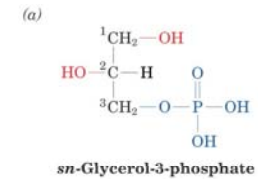
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The erythrocyte glycocalyx as revealed by electron microscopy using special staining techniques.

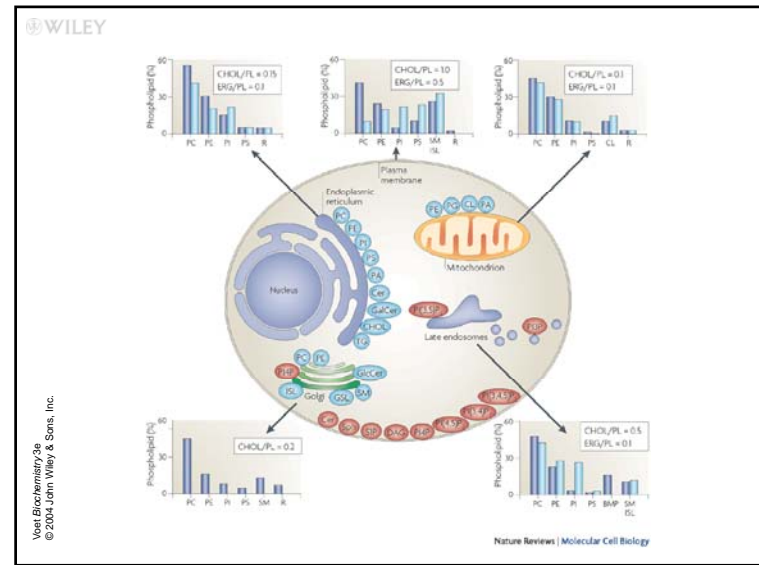
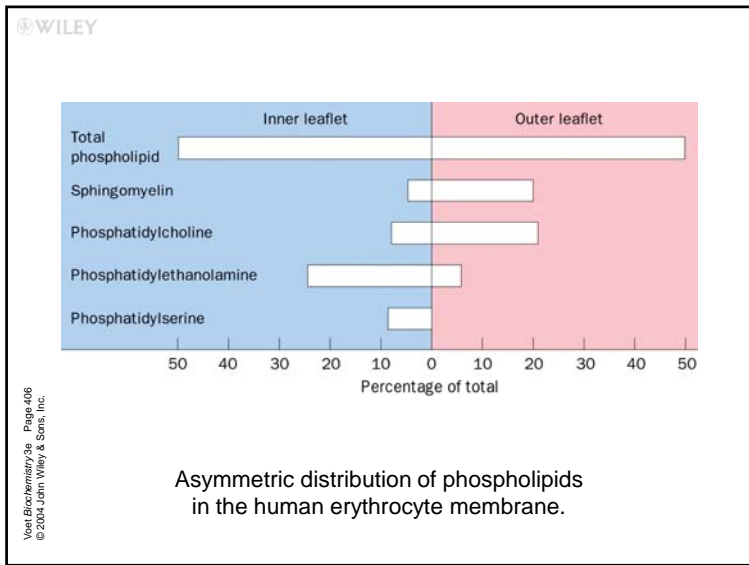
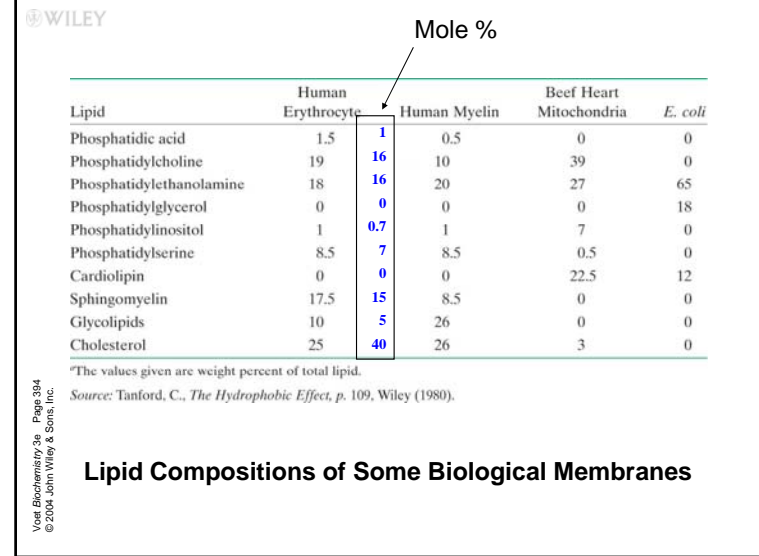
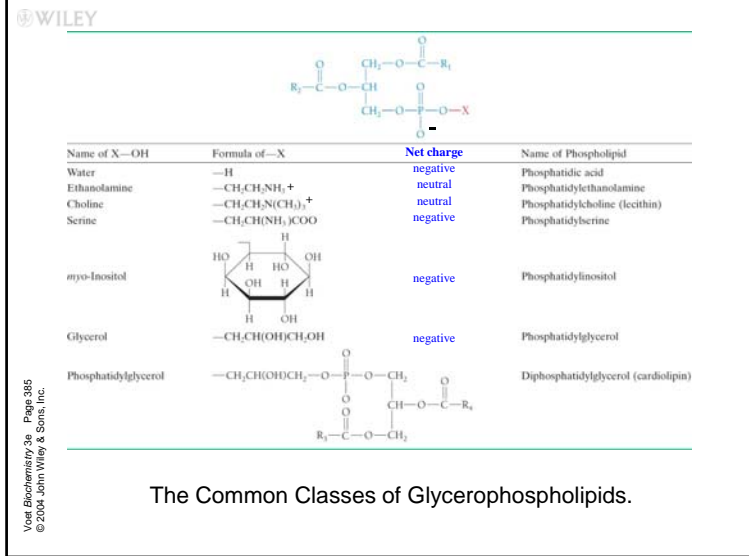
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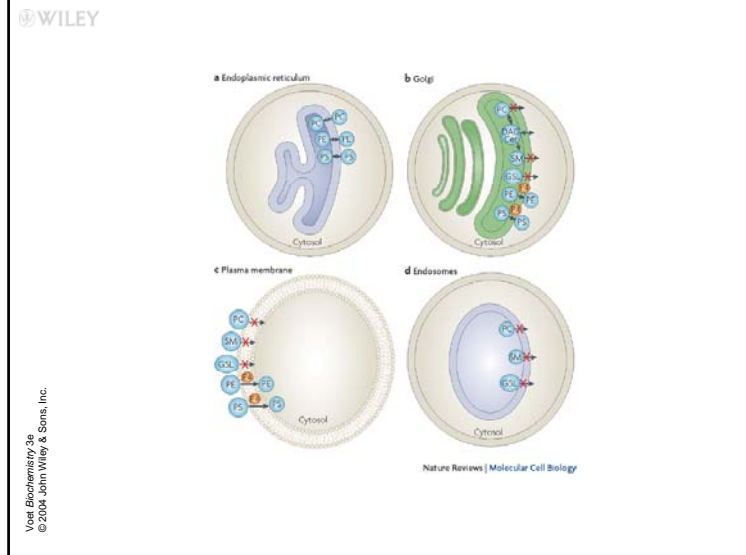
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Molecular formula of glycerophospholipids.

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DPPC = di palmitoyl phosphatidylcholine
DMPC = Can you guess?

Symbol*	Common Name	Systematic Name	Structure	mp (°C)
Saturated fatty acids				
12:0	Lauric acid	Dodecanoic acid	$\text{CH}_3(\text{CH}_2)_{10}\text{COOH}$	44.2
14:0	Myristic acid	Tetradecanoic acid	$\text{CH}_3(\text{CH}_2)_{12}\text{COOH}$	52
16:0	Palmitic acid	Hexadecanoic acid	$\text{CH}_3(\text{CH}_2)_{14}\text{COOH}$	63.1
18:0	Stearic acid	Octadecanoic acid	$\text{CH}_3(\text{CH}_2)_{16}\text{COOH}$	69.6
20:0	Arachidic acid	Eicosanoic acid	$\text{CH}_3(\text{CH}_2)_{18}\text{COOH}$	75.4
22:0	Behenic acid	Docosanoic acid	$\text{CH}_3(\text{CH}_2)_{20}\text{COOH}$	81
24:0	Lignoceric acid	Tetracosanoic acid	$\text{CH}_3(\text{CH}_2)_{22}\text{COOH}$	84.2

*Number of carbon atoms : number of double bonds. For unsaturated fatty acids, n is the number of carbon atoms, n - x is the double-bonded carbon atom, and x is the number of that carbon atom counting from the methyl terminal (ω) end of the chain.
 Source: Dawson, R.M.C., Elliott, D.C., Elliott, W.H., and Jones, K.M., *Data for Biochemical Research* (3rd ed.), Chapter 8, Clarendon Press (1986).

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The Common Biological Fatty Acids.

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Symbol*	Common Name	Systematic Name	Structure	mp (°C)
Unsaturated fatty acids (all double bonds are cis)				
16:1n-7	Palmitoleic acid	9-Hexadecenoic acid	$\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$	-0.5
18:1n-9	Oleic acid	9-Octadecenoic acid	$\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_9\text{COOH}$	13.4
18:2n-6	Linoleic acid	9,12-Octadecadienoic acid	$\text{CH}_3(\text{CH}_2)_4(\text{CH}=\text{CHCH}_2)_2(\text{CH}_2)_3\text{COOH}$	-9
18:3n-3	α-Linolenic acid	9,12,15-Octadecatrienoic acid	$\text{CH}_3(\text{CH}_2)_4(\text{CH}=\text{CHCH}_2)_3(\text{CH}_2)_3\text{COOH}$	-17
18:3n-6	γ-Linolenic acid	6,9,12-Octadecatrienoic acid	$\text{CH}_3(\text{CH}_2)_6(\text{CH}=\text{CHCH}_2)_3(\text{CH}_2)_3\text{COOH}$	-49.5
20:4n-4	Arachidonic acid	5,8,11,14-Eicosatetraenoic acid	$\text{CH}_3(\text{CH}_2)_4(\text{CH}=\text{CHCH}_2)_4(\text{CH}_2)_2\text{COOH}$	-54
20:5n-3	EPA	5,8,11,14,17-Eicosapentaenoic acid	$\text{CH}_3(\text{CH}_2)_3(\text{CH}=\text{CHCH}_2)_5(\text{CH}_2)_2\text{COOH}$	-54
22:6n-3	DHA	4,7,10,13,16,19-Docosahexaenoic acid	$\text{CH}_3(\text{CH}_2)_2(\text{CH}=\text{CHCH}_2)_6\text{COOH}$	39
24:1n-9	Nervonic acid	15-Tetracosenoic acid	$\text{CH}_3(\text{CH}_2)_9\text{CH}=\text{CH}(\text{CH}_2)_{13}\text{COOH}$	

*Number of carbon atoms : number of double bonds. For unsaturated fatty acids, n is the number of carbon atoms, n - x is the double-bonded carbon atom, and x is the number of that carbon atom counting from the methyl terminal (ω) end of the chain.
 Source: Dawson, R.M.C., Elliott, D.C., Elliott, W.H., and Jones, K.M., *Data for Biochemical Research* (3rd ed.), Chapter 8, Clarendon Press (1986).

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The Common Biological Fatty Acids.

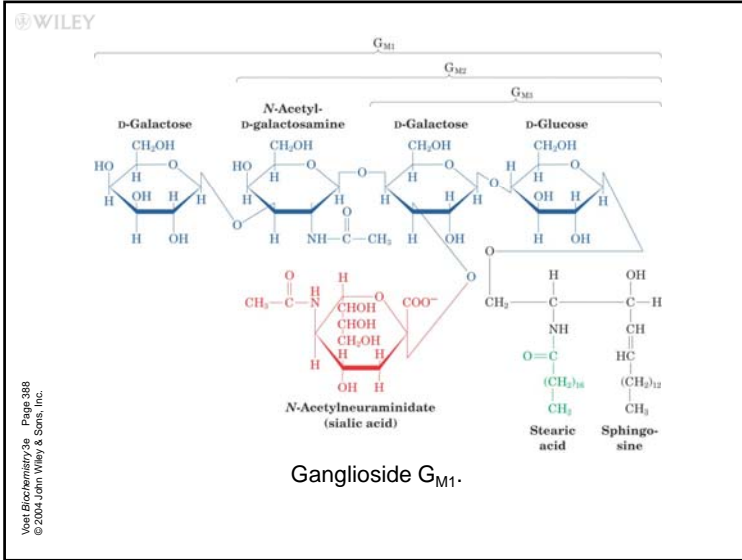
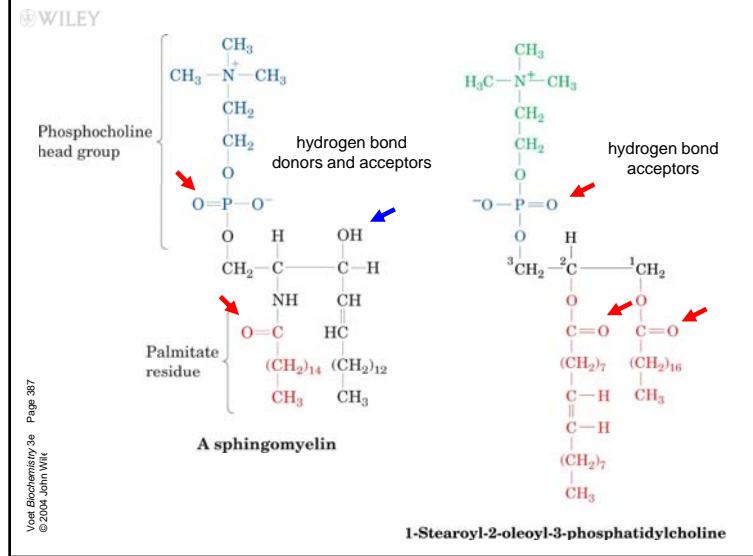
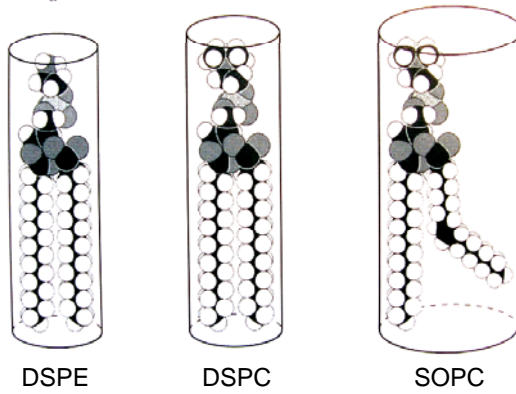
Phospholipids have different acyl chains

DCPC DLPC DMPC DPPC DSPC DAPC

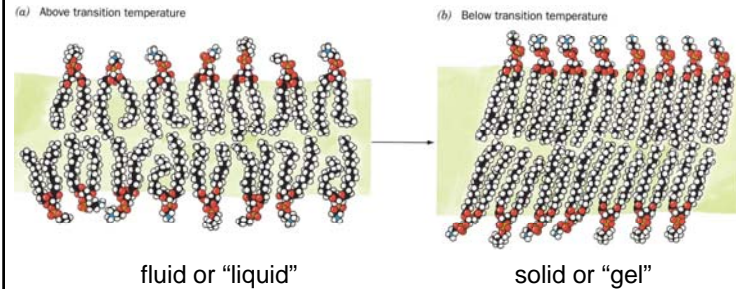
10 12 14 16 18 20

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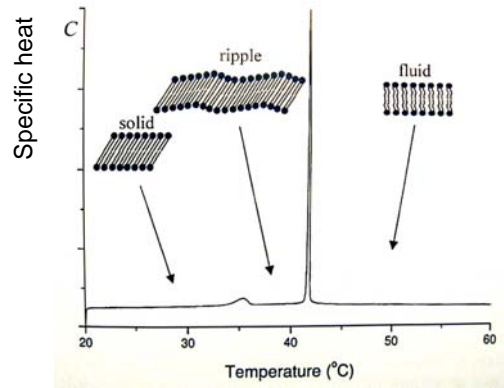
Dimensions of the phospholipids molecules



Exploring lateral heterogeneity in synthetic membranes



Phase transition and phases in DPPC



Phase diagram of lipid bilayers of binary mixtures of PC lipids

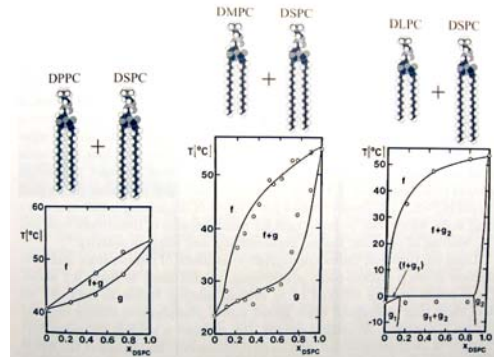
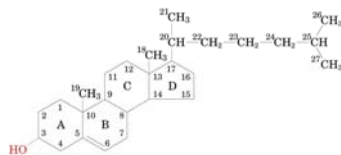


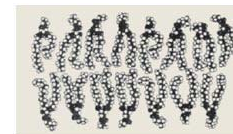
Fig. 9.7. Phase diagrams of lipid bilayers for three binary mixtures of PC lipids with different fatty-acid chain lengths. *f* denotes the liquid-disordered phase and *g* denotes solid-ordered phases

How does cholesterol affect the physical properties of lipids bilayers?

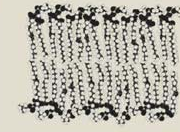


Nat. Rev. Mol. Cell Biol. 5, 389-399 (2004)
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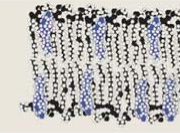
Illustration of Michael Peck, Ph.D., Bethesda, Maryland



Liquid-crystalline, liquid-disordered (L_d or L_d')
 $S = \text{Low}$
 $D_l = \text{Fast } (-1 \mu\text{m}^2 \text{s}^{-1})$



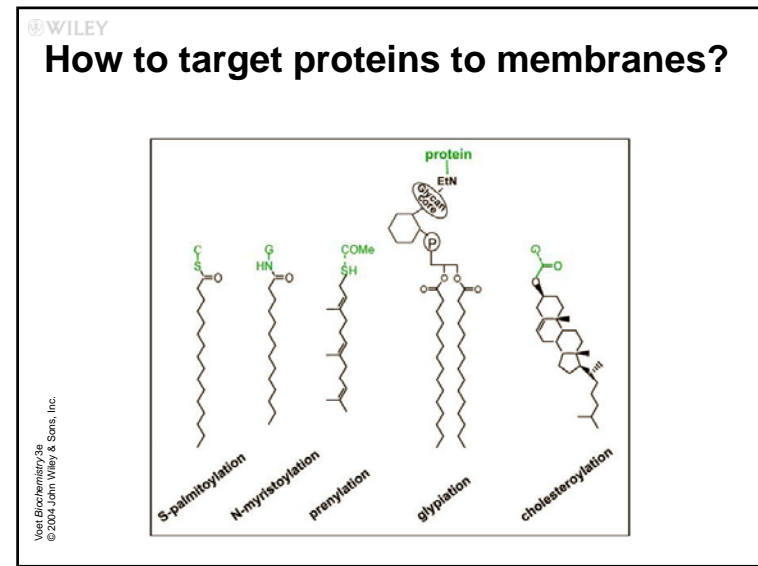
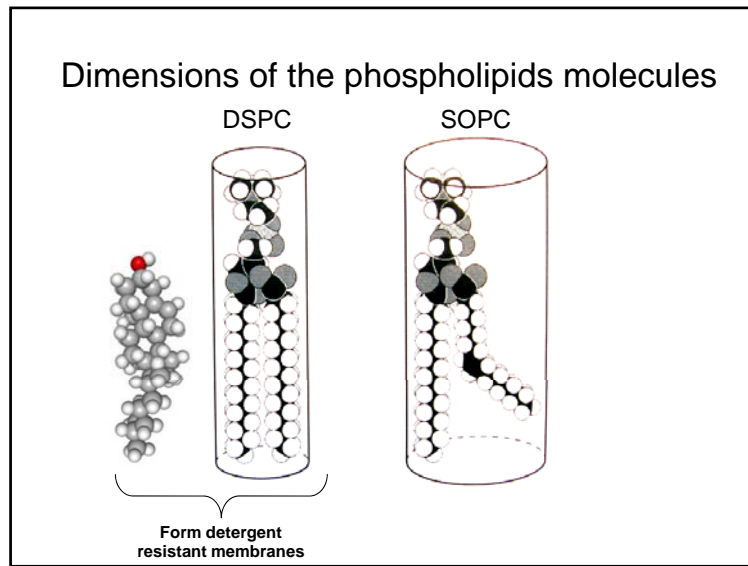
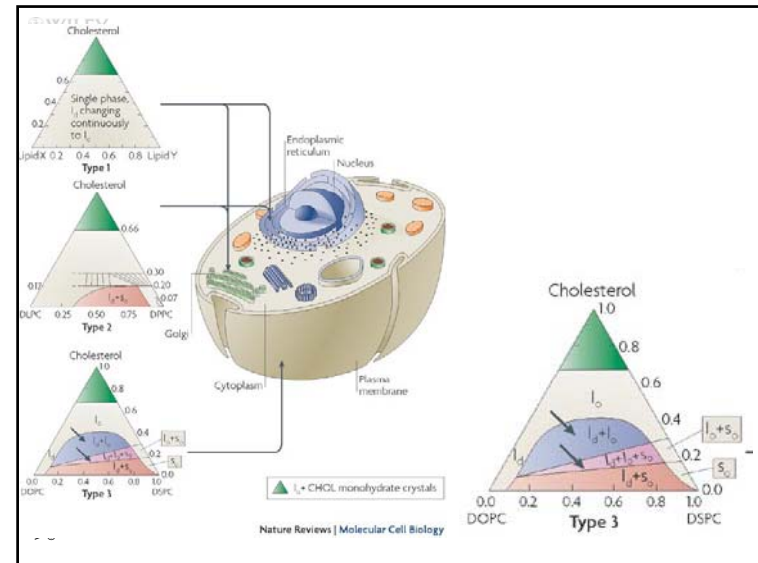
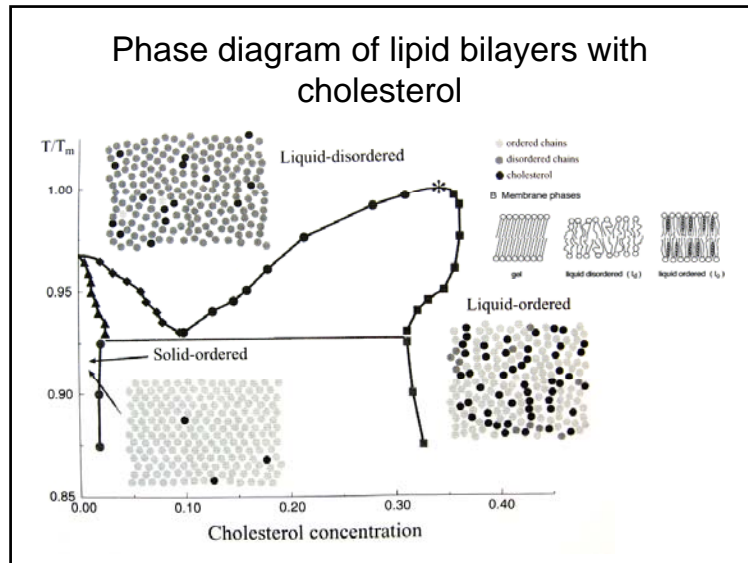
Solid gel (S_g or L_g)
 $S = \text{High}$
 $D_l = \text{Slow } (10^{-2} \mu\text{m}^2 \text{s}^{-1})$



Liquid-ordered, 'raft' (L_o or L_o')
 $S = \text{High}$
 $D_l = \text{Fast } (-1 \mu\text{m}^2 \text{s}^{-1})$

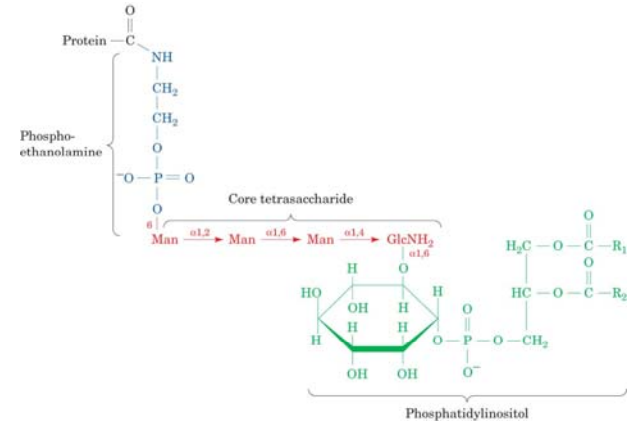
Nature Reviews | Molecular Cell Biology

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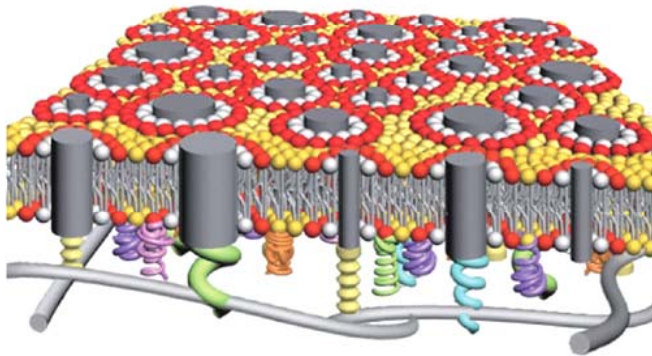
Covalent lipid modification of proteins

- Palmitoylation (16 C acyl chain)
 - (Cys) thioester bond
 - The most abundant
 - The only **reversible** (can be cleaved by thioesterases)
- N-myristoylation (14 C acyl chain)
 - In most cases not enough to anchor a protein to membrane
 - N-terminal Gly, amide bond
- Prenylation (farnesyl or geranylgeranyl)
 - Thioether linkage to a C-terminal Cys



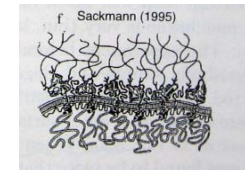
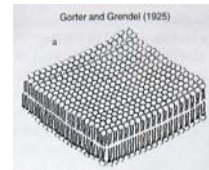
Core structure of the GPI anchors of proteins

Density of membrane proteins in the ER and Golgi
30000 per μm^2

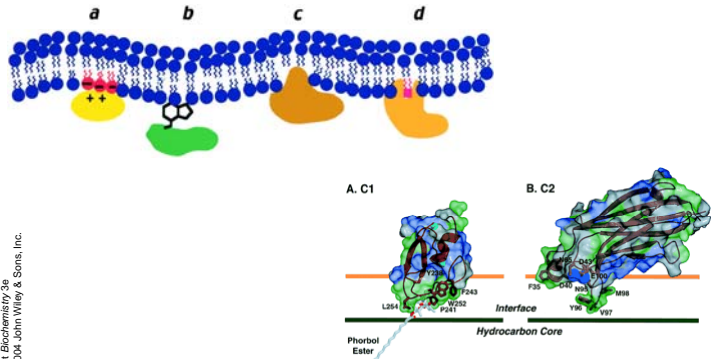


The membrane
is a protein-lipid composite
rather than a dilute solution
of protein in a lipid solvent

Biological membranes are different from synthetic membranes



Protein-lipid interactions



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Protein binding domains

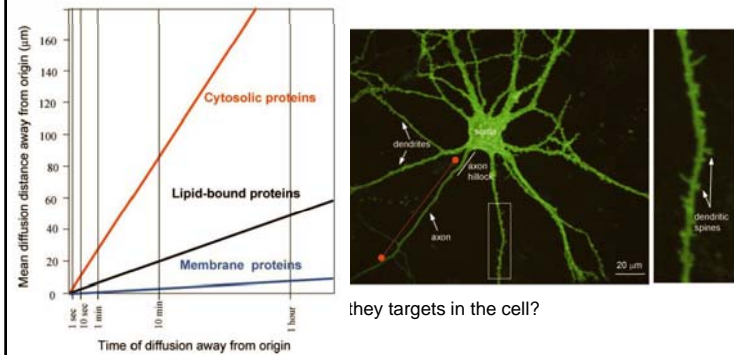
- Polybasic domains
 - Short cluster of basic amino acids (<30)
 - non-specific binding to acidic lipids
 - Act synergistically with lipid modifications
- Phospho inositides binding modules
 - Low affinity lipid interaction modules that regulate transient association w/cellular membranes
 - Examples:
 - PH, FERM, ENTH [PI(4,5)P₂]
 - FYVE, PX [PI(3)P]
- C1 and C2 domains

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Table 1. Occurrence of Membrane Targeting Motifs in the Human Genome

Membrane Targeting Motif	Number of Motifs in the Human Genome
Lipid-based:	
Palmitoylation	?
Myristoylation	270
Prenylation (CaaX)	214
Protein-based:	
Polybasic	?
PH domains	448
ENTH domains	16
FYVE domain	66
PX domains	65
C2 domain	225
C1 domain	97

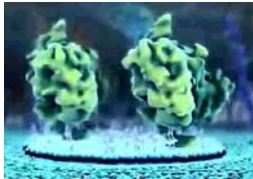
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they targets in the cell?

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What are lipid rafts?



Origin of the raft hypothesis

- glycosphingolipids cluster in Golgi before being sorted to the apical surface of polarized epithelial cell
 - Biochemistry 27, 6197 (1988)
- glycosphingolipids clusters are insoluble in Triton X-100 at 4 °C, and the detergent resistant membranes are rich in cholesterol, and GPI-anchored proteins
 - (1994)
- synthetic membranes composed of glycosphingolipids and cholesterol recapitulate the detergent-resistant characteristics of the glycosphingolipid clusters
 - (2001)

Two lines of inquiry emerged from the early raft studies

- exploring lateral heterogeneity in synthetic membranes
- addressing the functionality of glycosphingolipid-cholesterol rich domains in biological membranes

This characterizations have proven to be frustratingly difficult

Provisional contemporary definition of rafts

- **“Membrane rafts are small (10-200 nm), heterogeneous, highly dynamic, sterol- and sphingolipid-enriched domains that compartmentalize cellular processes”**

– 2006 Keystone Symposium

Suggested roles for Rafts

- Signal transduction pathways
- Apoptosis Cell adhesion and migration
- Synaptic transmission Organization of cytoskeleton
- Protein sorting during exo- and endocytosis
- Point of entry of viruses, bacteria, and toxins
- Site of viral assembly and formation of prion and Alzheimer amyloid

Conclusion

- The existence of lipid rafts **cannot** be taken as an established fact
- We should proceed with **caution** and **open mind** to alternative mechanisms for phenomena currently attributed to rafts

